CLAIMS

1. Process for preparing 2',3'-didehydro-2',3'-dideoxynucleosides of formula

5 P'O O B

$$\begin{array}{c}
O \\
B
\end{array}$$
(1)

in which

P' represents hydrogen or a suitable protecting group P, and

10 B represents a natural or modified, optionally substituted purine or pyrimidine base or a five- or six-membered monocyclic or eleven- or twelve-membered bicyclic, optionally substituted heterocyclic system containing at least one nitrogen atom;

which comprises the reductive elimination reaction of the compound of formula

P'O B (II)

in which

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X and Y represent, alternately, a halogen or an acyloxy group RCOO-,

P' and B have the meanings given above,

by reaction with zinc metal and a suitable activating agent,

characterized in that the divalent zinc is removed by precipitation, from an organic phase, of the corresponding zinc sulfide, by addition of a solution of an alkali metal or alkaline-earth metal sulfide to the said organic phase.

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2. Process according to Claim 1, in which:

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P' represents an acyl group RCO-, in which R represents a C_1 - C_5 alkyl R¹, preferably a methyl, or a group R¹COOC(R²R³)-, in which R¹, R² and R³ represent a C_1 - C_5 alkyl, preferably a methyl;

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- 5 B represents an optionally substituted natural purine or pyrimidine base, preferably adenine, inosine, 5-F-cytosine, hypoxanthine or thymine;
 - X and Y represent, alternatively, bromine and an acyloxy group RCOO-, in which R represents a C₁-C₅ alkyl R¹, preferably methyl, or a group R¹COOC(R²R³)-, in which R¹, R² and R³ represent a C₁-C₅ alkyl R¹, preferably a methyl.
 - Process according to Claim 1, in which the said activating agent is chosen from copper, acetic acid and ammonium or phosphonium salts, preferably ammonium or phosphonium salts.
 - 4. Process according to Claim 1, in which the said organic phase is chosen from solvents such as tetrahydrofuran, dimethylacetamide, alcohols, acetonitrile, chlorinated solvents and dimethyl sulfoxide, and mixtures thereof.
 - 5. Process according to Claim 1, in which the said sulfide solution comprises a polar solvent chosen from dipolar aprotic solvents and water, preferably water.
- 6. Process according to Claim 1, in which the said sulfide solution comprises the chosen alkali metal or alkaline-earth metal sulfide in an amount of at least one equivalent relative to the starting material, preferably in slight excess.
 - 7. Process according to Claim 1, in which the said mineral sulfide is an alkali metal or alkaline-earth metal sulfide, preferably sodium sulfide.
 - 8. Process according to Claim 1, in which the precipitated zinc sulfide is removed by filtration.
- 9. Process according to Claim 1, which further comprises the reduction reaction of the double bond of the compound of formula I to give the corresponding 2',3'-dideoxynucleoside of formula

in which X = Y = H, and P' and B have the meanings given above.

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10. Process according to Claim 1, which further comprises the deprotection reaction of a compound of formula

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in which P' represents a protecting group P, and B has the meanings given above,

to give the corresponding compound of formula I,

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in which P' represents hydrogen.

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11. Process according to Claim 9, which further comprises the deprotection reaction of a compound of formula

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in which P' represents a protecting group P, X and Y represent H, and B has the meanings given above,

to give the corresponding compound of formula II,

in which P' represents hydrogen.

5 12. Process for preparing 5-fluoro-2',3'-dideoxy-2',3'-didehydro-β -D-cytidine, stavudine, dideoxyadenosine, didanosine and zalcitabine, which comprises a process according to Claims 1 to 10.